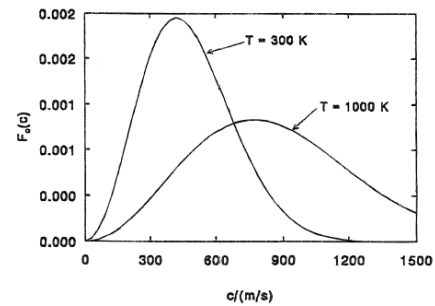


### Chapt 3 Heating and cooling

- ◆ Free electrons have a kinetic temperature, the only real temperature in the gas
- ◆ Heating is any process that gives energy to the gas, increasing the temperature
- ◆ Cooling is any process that removes energy from the gas, lowering the temperature
- ◆ Thermal equilibrium is when heating and cooling rates match
- ◆ Often radiation is the only heating & cooling

### A Maxwellian velocity distribution



For  $N_2$ , depends on mass <http://www.thermopedia.com/content/942>

### Thermal equilibrium

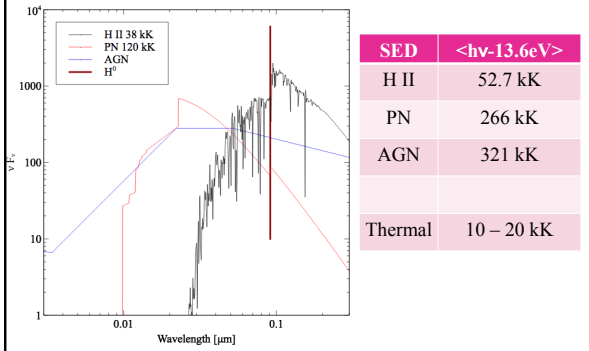
- ◆ Heating by radiation field in photo case
- ◆ In coronal case external process sets temperature
- ◆ Cooling is anything that converts kinetic energy into light that escapes

### Photoelectric heating

$$G(H) = n(H^0) \int_{\nu_0}^{\infty} \frac{4\pi J_\nu}{h\nu} h(\nu - \nu_0) a_\nu(H^0) d\nu \quad [\text{erg cm}^{-3} \text{s}^{-1}] \quad (3.1)$$

- ◆ Depends on SED shape

### SED, H<sup>0</sup> ion limit, photoelectron energy



### Ionization parameter

- ◆ Dimensionless ratio of hydrogen to ionizing photon densities

$$U = \frac{1}{4\pi r^2 c n_H} \int_{\nu_0}^{\infty} \frac{L_\nu}{h\nu} d\nu = \frac{Q(H^0)}{4\pi r^2 c n_H}, \quad (14.7)$$

$$n(X^{+i}) \int_{\nu_i}^{\infty} \frac{4\pi J_\nu}{h\nu} a_\nu(X^{+i}) d\nu = n(X^{+i}) \Gamma(X^{+i}) \quad (2.30)$$

$$= n(X^{+i+1}) n_e \alpha_G(X^{+i}, T),$$

### Photoelectric heating

- ◆ Heating proportional to photoionization rate, which is equal to  $n_e n_p \alpha$ , the recombination rate
- ◆ Heating depends on density squared

$$G(H) = n_e n_p \alpha_A(H^0, T) \frac{\int_{\nu_0}^{\infty} \frac{4\pi J_\nu}{h\nu} h(\nu - \nu_0) a_\nu(H^0) d\nu}{\int_{\nu_0}^{\infty} \frac{4\pi J_\nu}{h\nu} a_\nu(H^0) d\nu} \quad (3.2)$$

$$= n_e n_p \alpha_A(H^0, T) \frac{3}{2} kT_i$$

### Let's try different SEDs

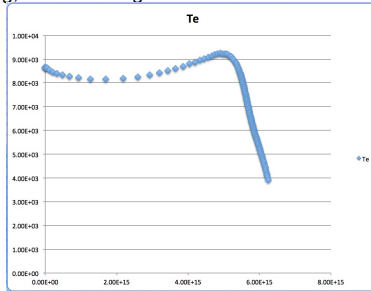
- ◆ Density 1 cm<sup>-3</sup>, constant temperature, one zone, same ionization parameter

## Photoelectric heating vs depth

### ◆ Dependence on depth

- Spectrum, heating, across H<sup>+</sup> region
- Yesterday's hiis.in
- Save continuum output

### ◆ Save heating



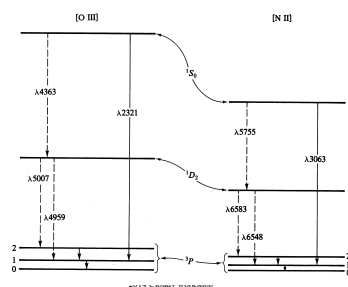
## Cooling

- ◆ Anything that converts kinetic energy (heat) into light (which escapes)
- ◆ Chap 3 lists a number of processes
- ◆ Collisional excitation of lines is normally the most important cooling process

$$L_C = n_e n_1 q_{12} h\nu_{21}. \quad (3.22)$$

## [O III]

### ◆ AGN3 Fig 3.1



## NIST

◆ <http://www.nist.gov/pml/data/asd.cfm>

Physical Measurement Laboratory

◆ <http://www.nist.gov/pml/data/asd.cfm>

NIST Atomic Spectra Database

Version 4

Welcome to the NIST Atomic Spectra Database, NIST Standard Reference Database #78. The spectroscopic data may be selected and displayed according to wavelengths or energy levels by choosing one of the following options:

- LINES** Spectral lines and associated energy levels displayed in wavelength order with all selected spectra intermixed or in multiples order. Transition probabilities for the lines are also displayed where available.
- LEVELS** Energy levels of a particular atom or ion displayed in order of energy above the ground state.

NIST ASD Team  
Principal Developers (Currently Active):  
Yu. Ralchenko, A. Kramida, and J. Reader

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2014 Clouby workshop

### NIST Atomic Spectra Database Levels Form

Best viewed with the latest versions of Web browsers and Java.

This form provides access to NIST critically evaluated data on atomic energy levels.

Spectrum:  e.g., Fe I

Default Values

Level Units:  Extended Search:  for all levels seen

Format output:

Display output:

Page size:

Term ordered:  term energy

Energy ordered:

Level information:  Principal configuration  Principal term  Level  J  Lande-g

### O III

Configuration	Term	J	Level (cm <sup>-1</sup> )
2s <sup>2</sup> 2p <sup>2</sup>	<sup>3</sup> P	0	0
		1	113.178
		2	306.174
2s <sup>2</sup> 2p <sup>2</sup>	<sup>1</sup> D	2	20 273.27
2s <sup>2</sup> 2p <sup>2</sup>	<sup>1</sup> S	0	43 185.74
2s2p <sup>3</sup>	<sup>5</sup> S <sup>o</sup>	2	60 324.79
2s2p <sup>3</sup>	<sup>3</sup> D <sup>o</sup>	3	120 025.2
		2	120 053.4
		1	120 058.2

## Heating – cooling balance

- ◆ Both heating and cooling depend on square of density
- ◆ So no density dependence
- ◆ Try it! Remove constant temperature command, compare temperatures at two densities

## Other cooling processes

- ◆ Save cooling command
- ◆ Look at various output

## Coronal equilibrium

- ◆ Mechanical energy sets kinetic temperature
- ◆ “Coronal” command in Cloudy
- ◆ Try several T, plot SAVE CONTINUUM output

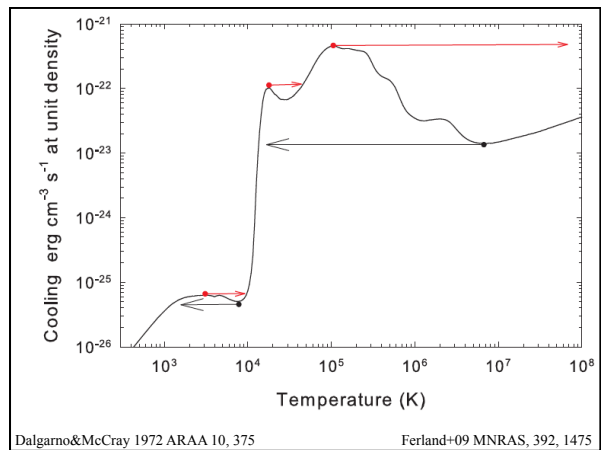


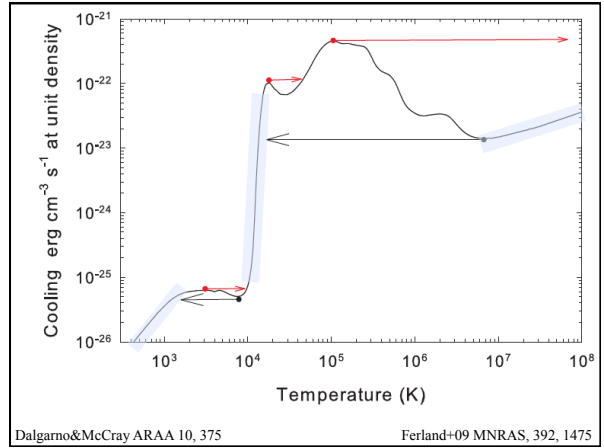
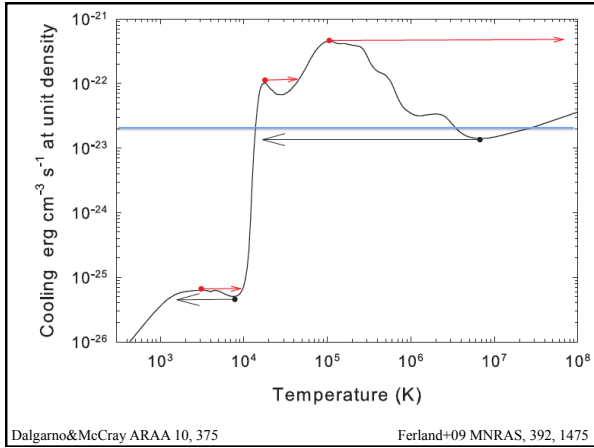
## Try different temperatures

- ◆ Coronal command
- ◆ Unit cell
- ◆ Plot spectrum
- ◆ Must include “cosmic ray background”

## Grid command – cooling function

- ◆ Grid command Hazy 1 Chapter 18
  - Carefully study temperature log rules, Sec 18.5
- ◆ Coronal equilibrium command
- ◆ Save cooling output
- ◆ Plot cooling vs temperature





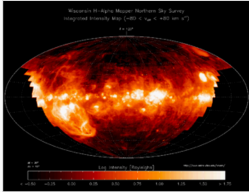
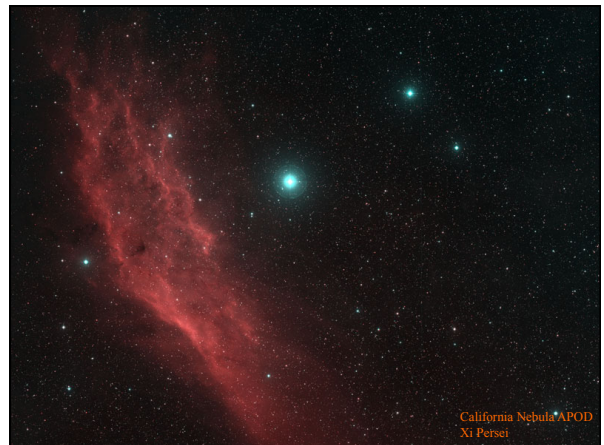
[http://en.wikipedia.org/wiki/Interstellar\\_medium](http://en.wikipedia.org/wiki/Interstellar_medium)

### Interstellar medium

From Wikipedia, the free encyclopedia

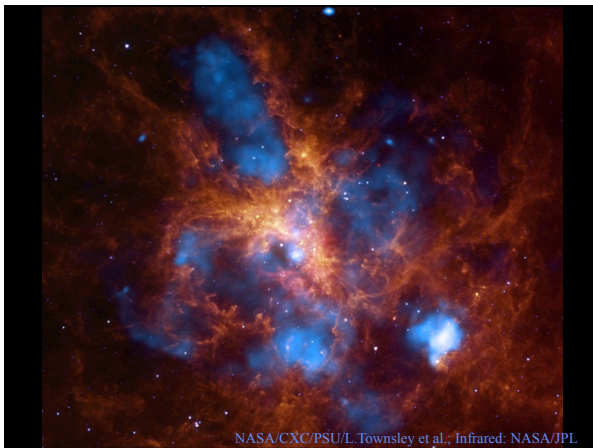
*For other uses, see Interstellar (disambiguation).*

In astronomy, the **interstellar medium** (or **ISM**) is the matter that exists in the space between the star systems in a galaxy. This matter includes gas in ionic, atomic, and molecular form, dust, and cosmic rays. It fills interstellar space and blends smoothly into the surrounding intergalactic space. The energy that occupies the same volume, in the form of electromagnetic radiation, is the **interstellar radiation field**.

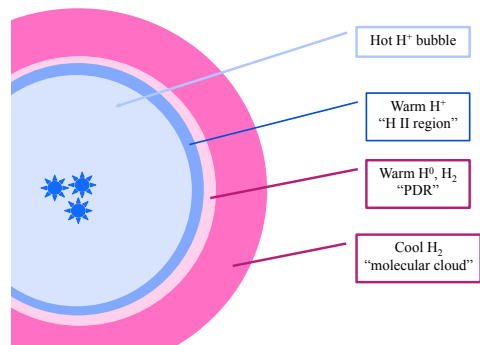



### Star forming H II regions

- ◆ Hot young stars very close to the molecular cloud that formed it
- ◆ Ionizing radiation and stellar winds strike nearby molecular cloud



### Idealized structure of an H II region



## Make spectra of stable phases

- ◆ Cold, warm, hot stable phases
- ◆ Ccurve.in
  - Remove grid, vary option
  - Leave ISM abundances
  - Save continuum (units microns), cooling
- ◆ Compute stable points
  - T=5e2K 2e4K, 8e4K, 1.5e6K, 2e7K

## Effects of U on ionization, temperature, & spectrum

- ◆ Let's use
  - A) an AGN SED
  - B) a low density, hden 0
  - C) unit volume
  - D) solar abundances
  - E) save the emitted continuum
  - F) and vary U;  $-5 \leq U \leq 3$
- ◆ Plot emitted continuum, 1e-4 to 1e3 microns, y axis 1e-20 down to 1e-26
- ◆ Temperature, peak ionization of Fe

## “make” parallel

- ◆ <https://trac.nublado.org/wiki/MpiParallel>

## Vary Metals – constant temperature



**Vary Metals –temperature balance**

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**Three-phase pressure stability**

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◆ `tsuite / auto / ism_grid`

**Vary blackbody temperature**

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